

9th ITG International Vacuum Electronics Workshop 2024

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# Removal of Persistent Micropollutants From Wastewater By Hybrid Treatment With Low-Energy Electrons and Ozone

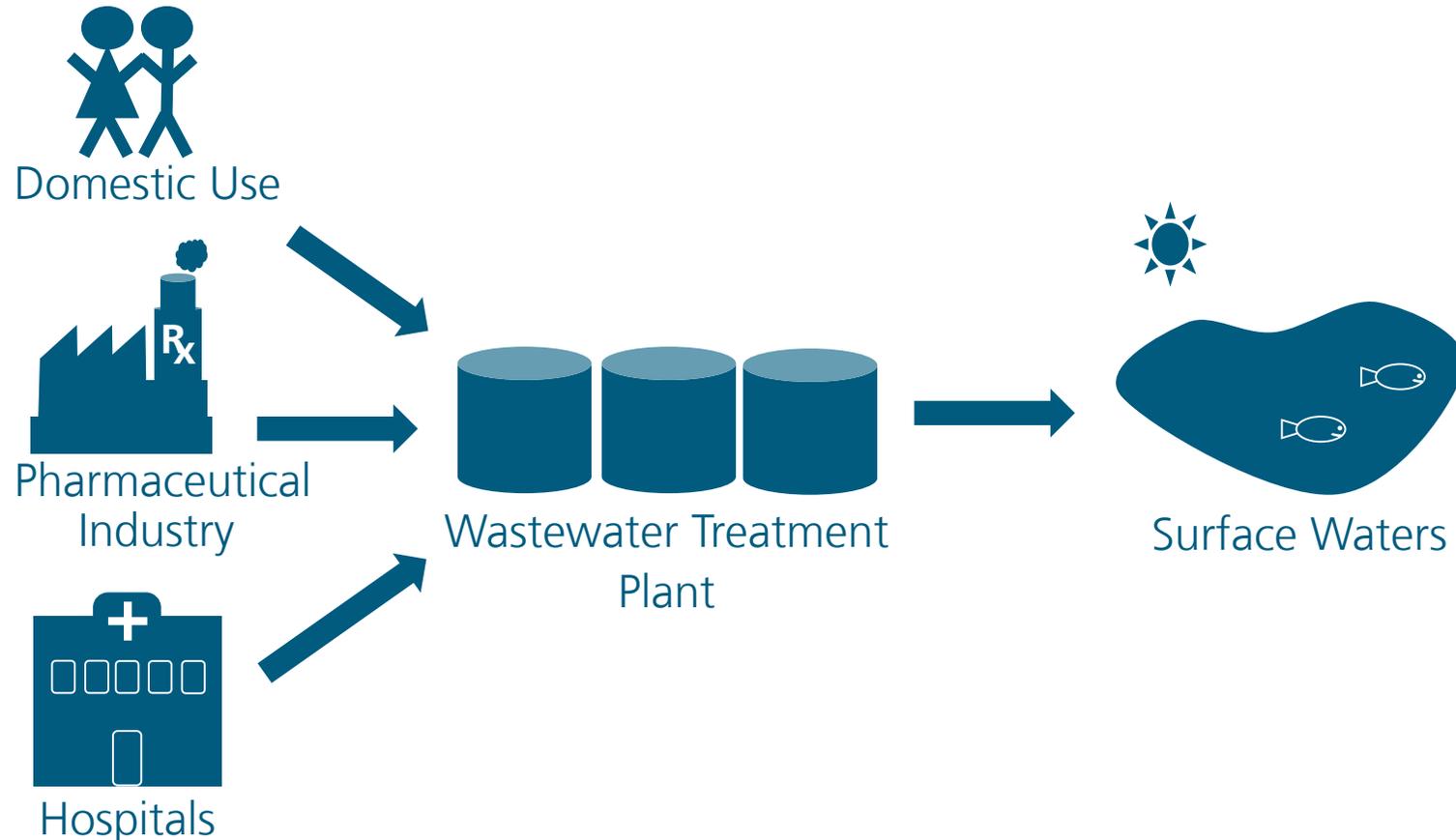
Lotte Ligaya Schaap, Tobias Teichmann, Andre Poremba, Gösta Mattausch, Simone Schopf,  
Elizabeth von Hauff

## AGENDA

- Introduction
- Methodology
- Results
- Work in Progress
- Conclusion and Outlook

# Introduction

## Pharmaceuticals in the Aquatic Environment

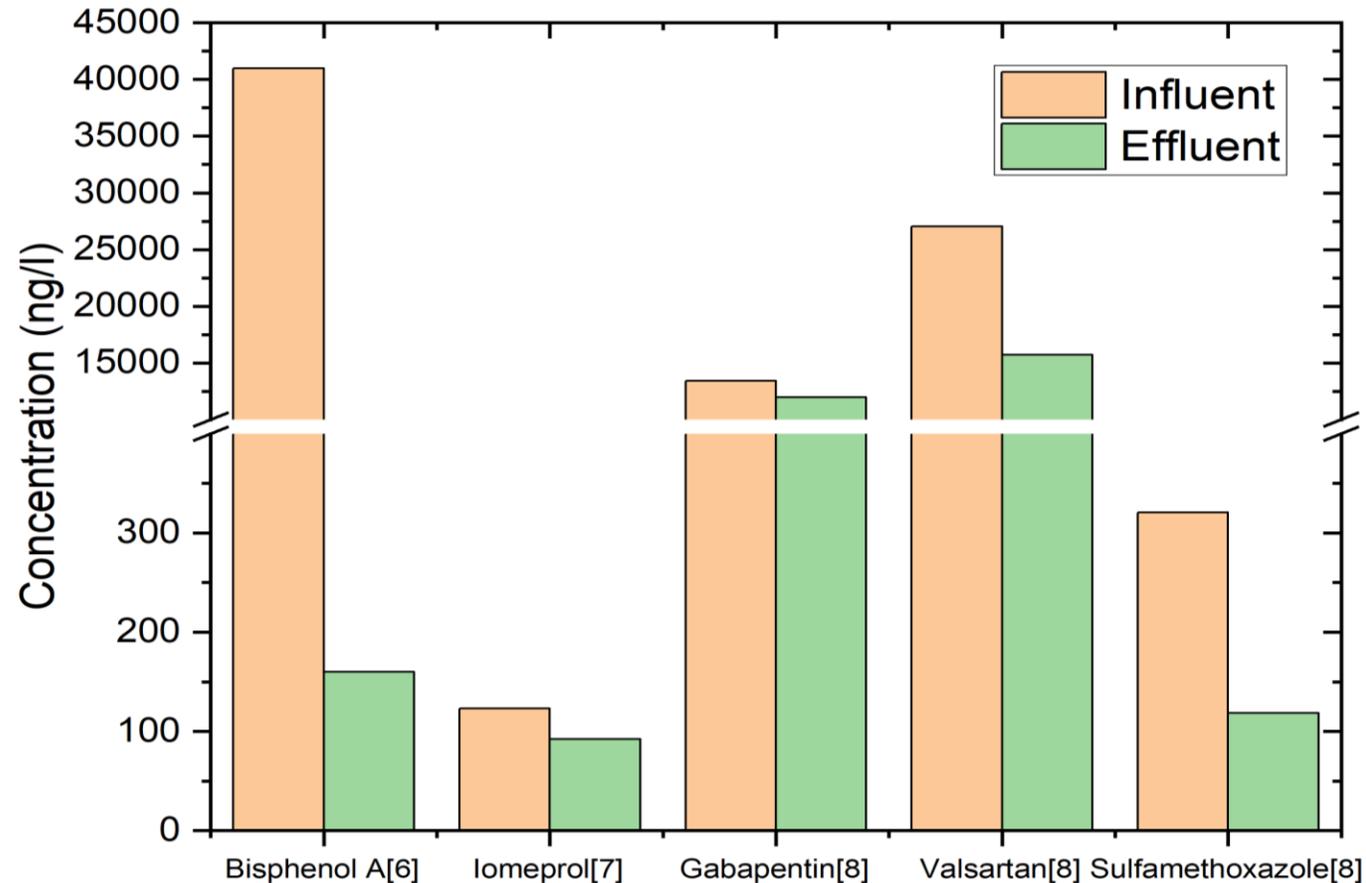


- In Germany alone, **9,600 to 12,800 tons of unused pharmaceuticals** are disposed of or flushed down the toilet yearly [1]
- Their presence in effluents have been linked to **ecological toxicity** [2], **antimicrobial resistance** [3], **endocrine disruption**, and **gradual reduction of fertility** [4,5]

**Figure 1.** Main sources of pharmaceuticals in the aquatic environment.

# Introduction

## Pharmaceuticals in the Aquatic Environment



**Figure 2.** Concentration of micropollutants before and after conventional wastewater treatment.



**The Problem:**

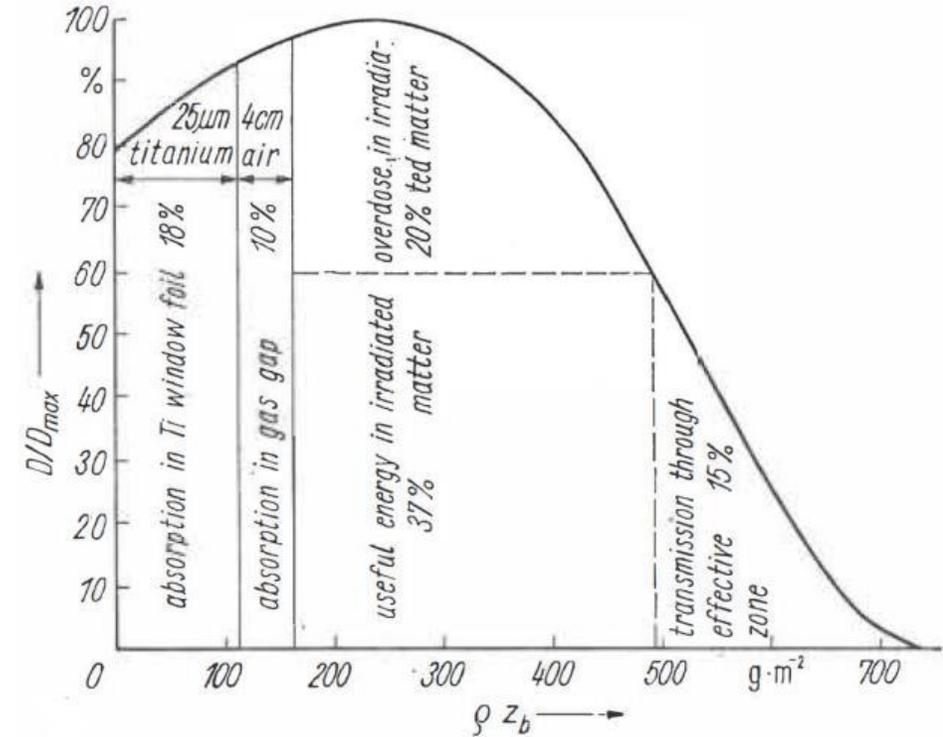
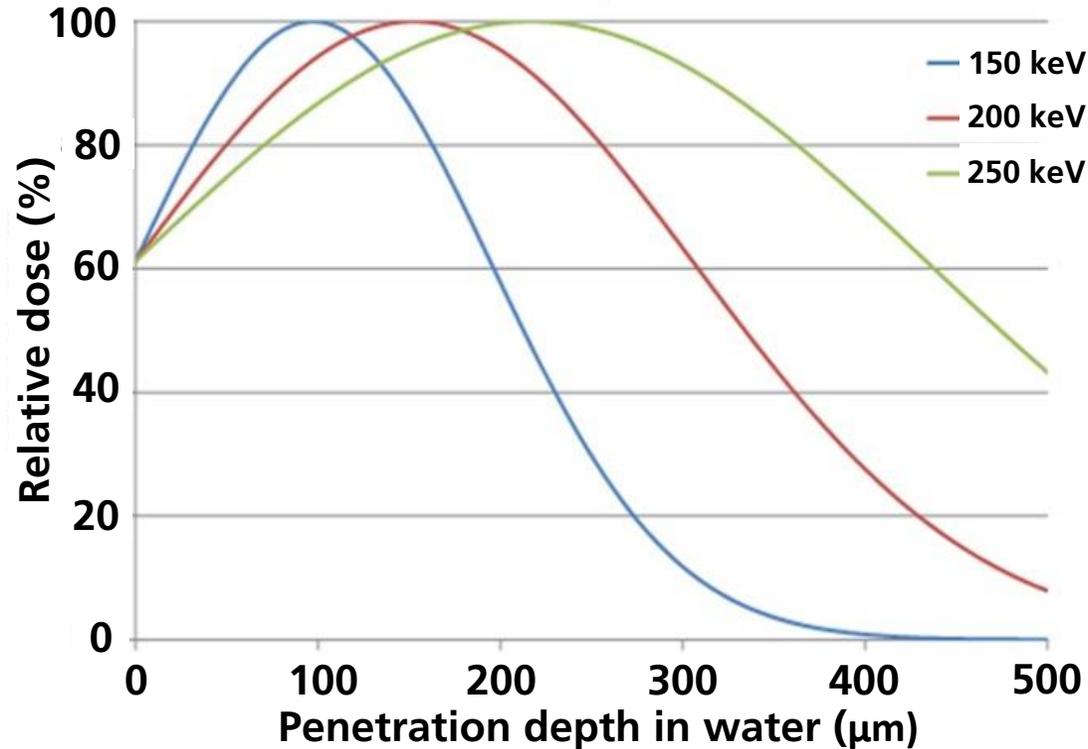
Pollutants of low-level toxicity which cannot be degraded by traditional methods

**The Solution:**

Low energy EB Treatment

# Introduction

## The Problem with Low-Energy Electrons

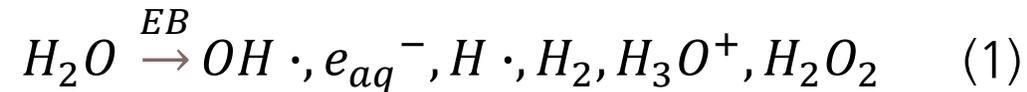


**Figure 3.** Penetration depth of accelerated electrons according to their energy and density of the absorber (left). Window foil and separating gap are sources of losses in low-energy EB processes (right) [9].

# Introduction

## Hybrid Wastewater Treatment

The irradiation of water produces several highly reactive, oxidizing and reducing radical species [10]:



**By combining low energy EB and ozone:**



The demands on radiation protection are kept to a minimum



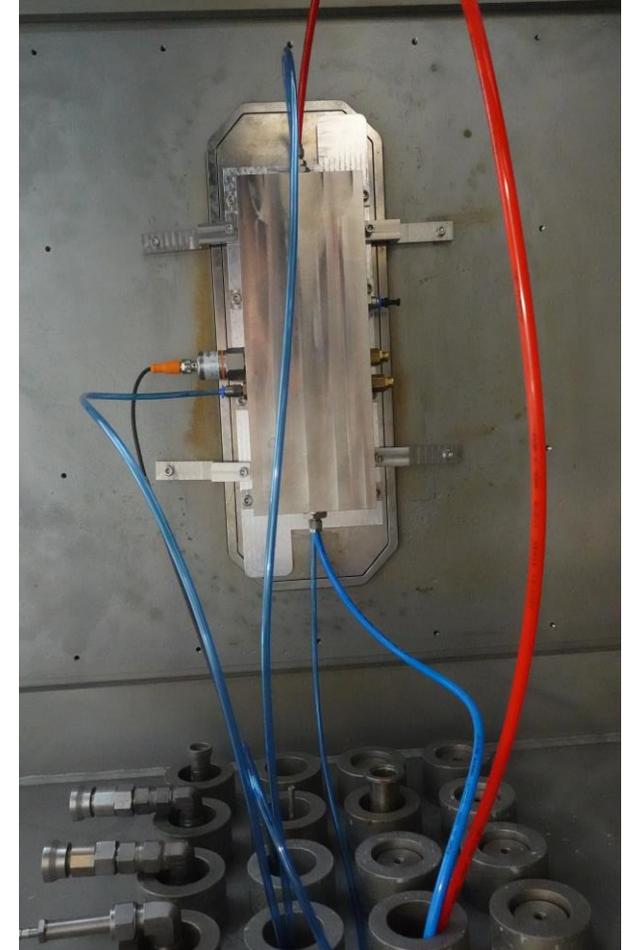
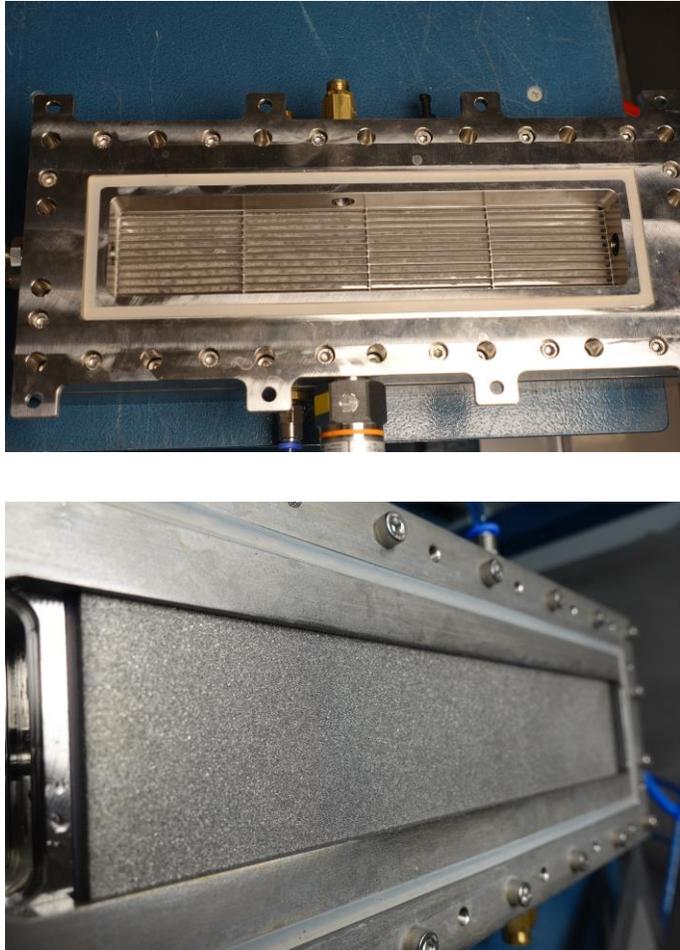
Low EB source dimensions and costs



An economic, flexible and compact treatment module is realized

# Methodology

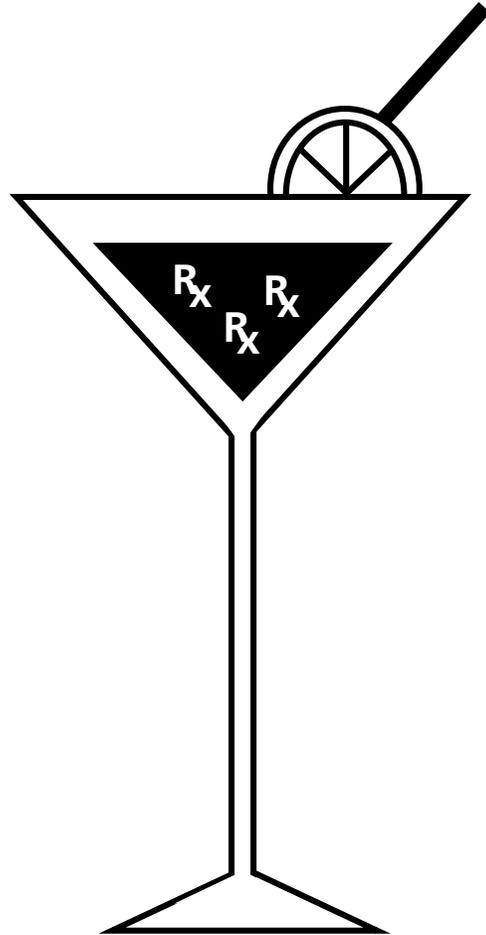
## Experimental Device



**Figure 4.** Treatment module (left) and treatment module inside REAMODE (right).

# Methodology

## Substance Irradiation



Pharmaceutical "cocktail" of five select substances in *deionized (DI) water and synthetic wastewater* were mixed



Cocktail was treated using treatment module



Concentration of substances before and after treatment determined using high-performance liquid chromatography

# Methodology

## Substance Irradiation



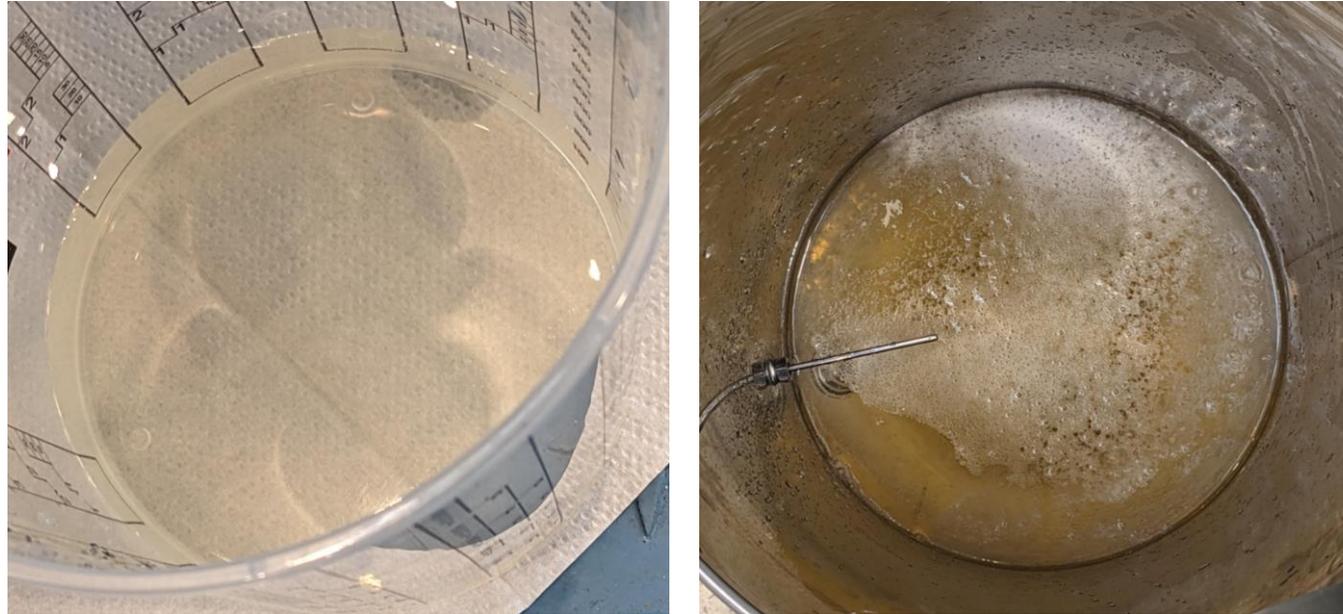
**Figure 5.** EB Facility REAMODE.

Irradiation done at Fraunhofer FEP's electron beam facility REAMODE with the following parameters:

- Maximum electron energy 200 keV
- Irradiation current 0.1-4 mA and 2 mA (multiple passes)
- Volume flow 4 l/min
- Gas flow 2 l/min

# Results

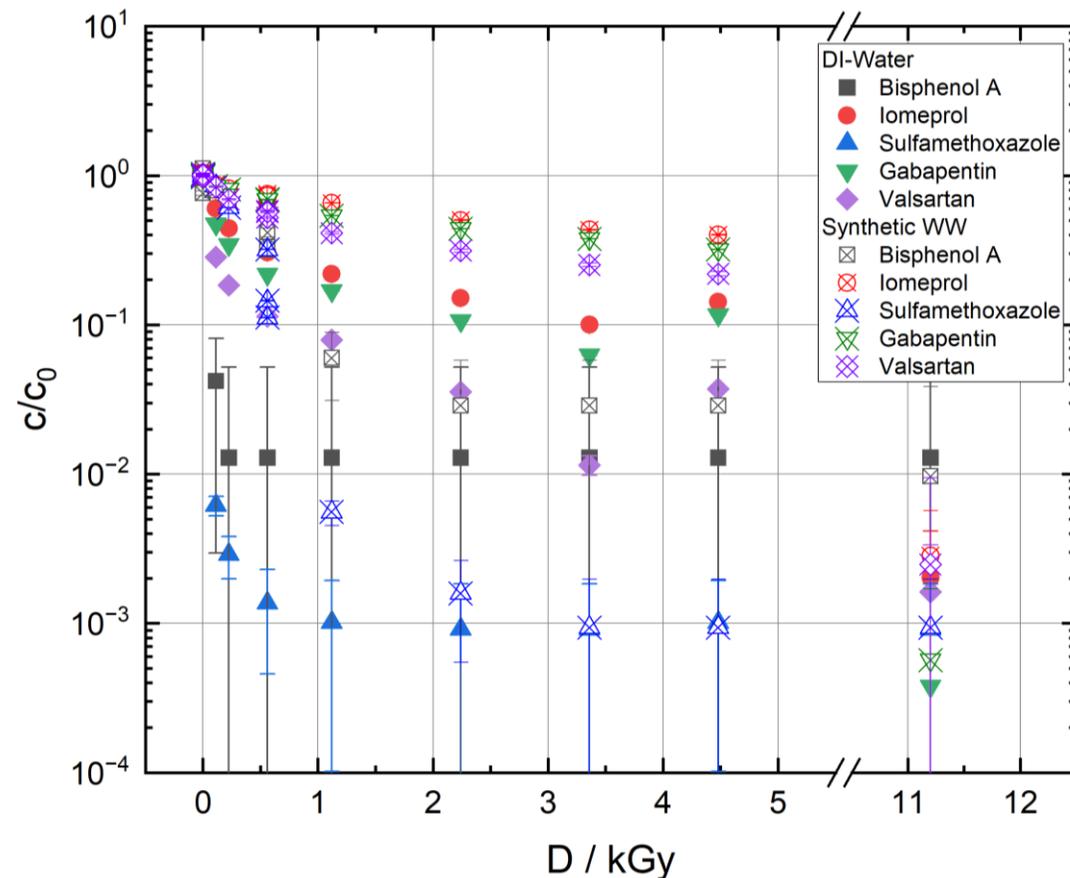
## Substance Irradiation



**Figure 6.** Pharmaceutical cocktail after irradiation.

# Results

## Substance Irradiation



**Significant degradation (>80%) achieved already at 2.2 kGy for all substances in deionized water and synthetic wastewater.**

**Figure 7.** Relative concentration vs dose for five substances in DI-water and synthetic wastewater.

# Results

## Substance Irradiation

**Table 1.** Doses at which significant degradation (>80%) is achieved for five select substances in deionised water (DI) and synthetic wastewater (SW).

Substance	Deionised Water	Synthetic Wastewater
Bisphenol A	0.1 kGy	0.6 kGy
Sulfamethoxazole	0.1 kGy	0.6 kGy
Valsartan	0.2 kGy	1.1 kGy
Gabapentin	1.1 kGy	1.1 kGy
lomeprol	2.2 kGy	2.2 kGy

# Results

## Comparing Wastewater Treatment Methods

### Energy Demands of Different Wastewater Treatment Methods

#### **Powdered Organic Carbon (PAC)**

**10-40 mg/l**

- 0.020-0.035 kWh/m<sup>3</sup> for adsorption step [11]
- 0.45 kWh/m<sup>3</sup> for whole process [12]

#### **Ultraviolet (UV) Treatment**

**7200 J/m<sup>2</sup>**

- 0.026 kWh/m<sup>3</sup> for irradiation step [13]
- 0.5-1 kWh/m<sup>3</sup> for whole process [12]

#### **EB + Ozone**

**2.2 kGy**

- 0.003 kWh/m<sup>3</sup> for irradiation step

# Results

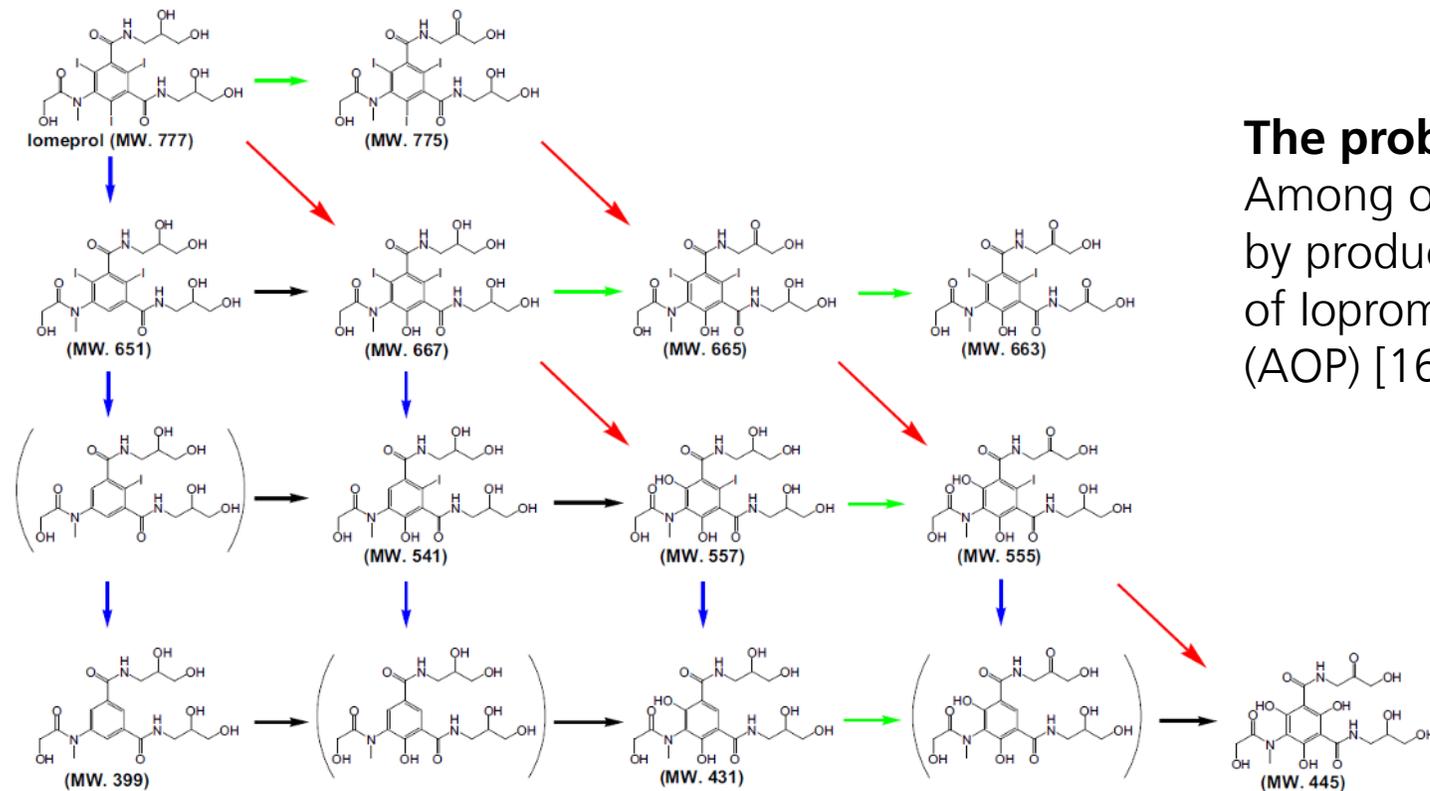
## Comparing Wastewater Treatment Methods

**Table 2.** Measured removal efficiencies with EB and ozone for 5 select substances in synthetic wastewater compared to traditional methods: powdered organic carbon (PAC) and ultra-violet (UV).

Substance	EB + O <sub>3</sub> (2.2 kGy)	EB + O <sub>3</sub> (11 kGy)	PAC [14]	UV [14]
Bisphenol A	97.1%	> 99.0%	5.3%	67.0%
lomeprol	49.7%	> 99.7%	65.0%	65.0%
Sulfamethoxazole	99.8%	> 99.9%	31.0%	81.0%
Gabapentin	56.0%	> 99.9%	41.0%	2.0%
Valsartan	68.6%	> 99.8%	99.0%	5.0%

# Work in Progress

## Testing Biodegradability

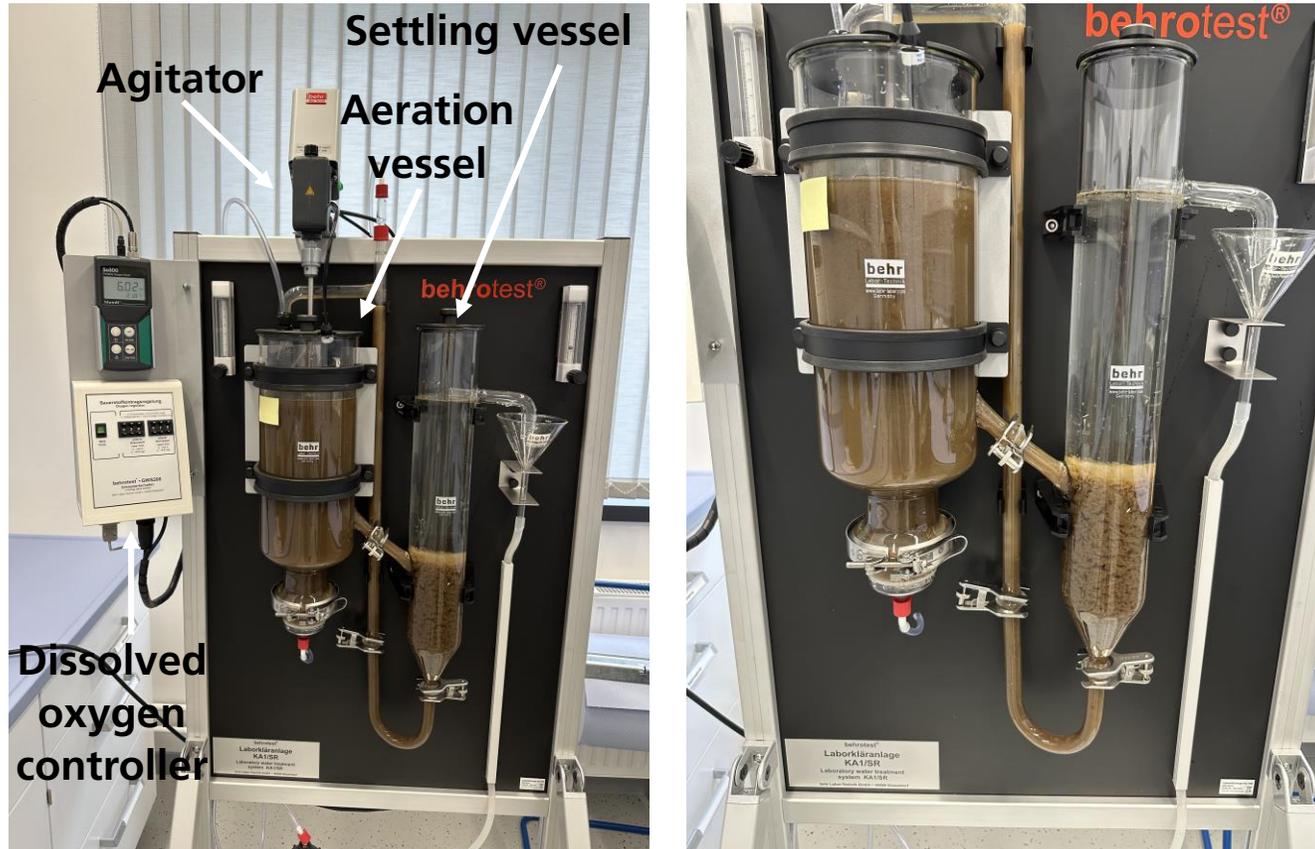


**The problem with degradation by-products:** Among other examples, 30 different degradation by products are formed from the transformation of lopromide by advanced oxidation processes (AOP) [16].

**Figure 8.** Degradation pathway of lopromide. Taken from Jeong et al. [15].

# Work in Progress

## Testing Biodegradability



**Figure 9.** Laboratory scale wastewater treatment plant.

### Meet Daisy, FEP's Laborkläranlage!

- The presence of numerous degradation by-products highlights the need for a biodegradability test
- Degradability is measured by taking chemical oxygen demand (COD) measurements
- 70% degradation = inherent ultimate biodegradability

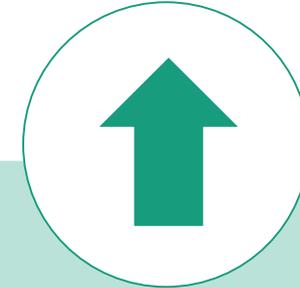
# Conclusion and Outlook



The hybrid treatment of EB + Ozone successfully degraded five select pharmaceuticals in DI water and synthetic wastewater



Analysis of degradation byproducts and their fate in further wastewater treatment and the environment is needed



An upscaling of the concept and expansion of its field of application must be developed

# Thank you for your attention !



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Beam and Plasma Technology FEP

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# Appendix

**Table 3.** Removal efficiencies for 5 select substances in deionised water (DI) and synthetic wastewater (SW).

Substance	Medium	RE (2,2 kGy)	RE (11 kGy)
Bisphenol A	DI	> 98,7%	> 98,7%
	SW	97,1%	> 99,0%
Iomeprol	DI	84,8%	> 99,8%
	SW	49,7%	> 99,7%
Sulfamethoxazole	DI	> 99,9%	> 99,9%
	SW	99,8%	> 99,9%
Gabapentin	DI	89,3%	> 99,9%
	SW	56,0%	> 99,9%
Valsartan	DI	96,4%	> 99,8%
	SW	68,6%	> 99,8%

# References

- [1] T. J. Scheytt, P. Mersmann, and T. Heberer, "Mobility of pharmaceuticals carbamazepine, diclofenac, ibuprofen, and propyphenazone in miscible-displacement experiments," *Journal of contaminant hydrology*, **83**, 1-2 (2006); doi: 10.1016/j.jconhyd.2005.11.002.
- [2] L. Szabó *et al.*, "Electron beam treatment for tackling the escalating problems of antibiotic resistance: Eliminating the antimicrobial activity of wastewater matrices originating from erythromycin," *Chemical Engineering Journal*, **321** (2017); doi: 10.1016/j.cej.2017.03.114.
- [3] A. Capodaglio, "Contaminants of Emerging Concern Removal by High-Energy Oxidation-Reduction Processes: State of the Art," *Applied Sciences*, **9**, 21 (2019); doi: 10.3390/app9214562.
- [4] H. M. Coleman, M. I. Abdullah, B. R. Eggins, and F. L. Palmer, "Photocatalytic degradation of 17 $\beta$ -oestradiol, oestriol and 17 $\alpha$ -ethinyloestradiol in water monitored using fluorescence spectroscopy," *Applied Catalysis B: Environmental*, **55**, 1 (2005); doi: 10.1016/j.apcatb.2004.07.004.
- [5] A. Kimura, M. Taguchi, Y. Ohtani, Y. Shimada, H. Hiratsuka, and T. Kojima, "Treatment of wastewater having estrogen activity by ionizing radiation," *Radiation Physics and Chemistry*, **76**, 4 (2007); doi: 10.1016/j.radphyschem.2006.04.005.
- [6] Tappert, L., Bunge, M., Hoehne, D., Dlugi, I., Fetters, K., Fischer, B., ... & Gestermann, S. (2024). Bisphenol A in surface waters in Germany: Part I. Reassessment of sources and emissions pathways for FlowEQ modeling. *Integrated Environmental Assessment and Management*, 20(1), 211-225.
- [7] Kormos, J. L., Schulz, M., & Ternes, T. A. (2011). Occurrence of iodinated X-ray contrast media and their biotransformation products in the urban water cycle. *Environmental science & technology*, 45(20), 8723-8732.

# References

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- [8] Patel, M., Kumar, R., Kishor, K., Mlsna, T., Pittman Jr, C. U., & Mohan, D. (2019). Pharmaceuticals of emerging concern in aquatic systems: chemistry, occurrence, effects, and removal methods. *Chemical reviews*, 119(6), 3510-3673.
- [9] S. Schiller et al., *Electron Beam Technology*, 477 (1982).
- [10] H. Bluhm, B. Han, A. G. Chmielewski *et al.*, "Electron beam devices for materials processing and analysis," in *Vacuum Electronics – Components and Devices*, J. A. Eichmeier and M. Thumm, Eds., Springer Science & Business Media, Berlin, Heidelberg, New York (2008).
- [11] Mousel, D., Kребber, K., Palmowski, L., Lyko, S., Thöle, D., & Pinnekamp, J. (2015). Energy demand for micropollutant removal in municipal wastewater treatment plants. IWA YWP Advanced wastewater treatment and water reuse—the future is now.
- [12] Adamczak, K., Lyko, S., Nafó, I., Evenblij, H., Cornelissen, A., Igos, E., ... & Stalder, T. (2012). Pharmaceutical residues in the aquatic system—a challenge for the future.
- [13] Bailey, J. R., Ahmad, S., & Batista, J. R. (2020). The impact of advanced treatment technologies on the energy use in satellite water reuse plants. *Water*, 12(2), 366.
- [14] Verlicchi Aukidy , M.; , P.; Al and treatment of hospital effluent? 514, pp. 467 Zambello , E. (2015): What have we learned from worldwide experiences on the management an overview and a discussion on perspectives. In *The Science of the total environment* 491. DOI: 10.1016/j.scitotenv.2015.02.020.

# References

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[15] Jeong, J., Jung, J., Cooper, W. J., & Song, W. (2010). Degradation mechanisms and kinetic studies for the treatment of X ray contrast media compounds by advanced oxidation/reduction processes. *Water Research*, 44(15), 4391

[16] Nowak, A., Páček, G., & Mroziček, A. (2020). Transformation and ecotoxicological effects of iodinated X-ray contrast media. *Reviews in Environmental Science and Bio/Technology*, 19(2), 337-354