

9th ITG International Vacuum Electronics Workshop 2024

Removal of Persistent Micropollutants From Wastewater By Hybrid Treatment With Low-Energy Electrons and Ozone

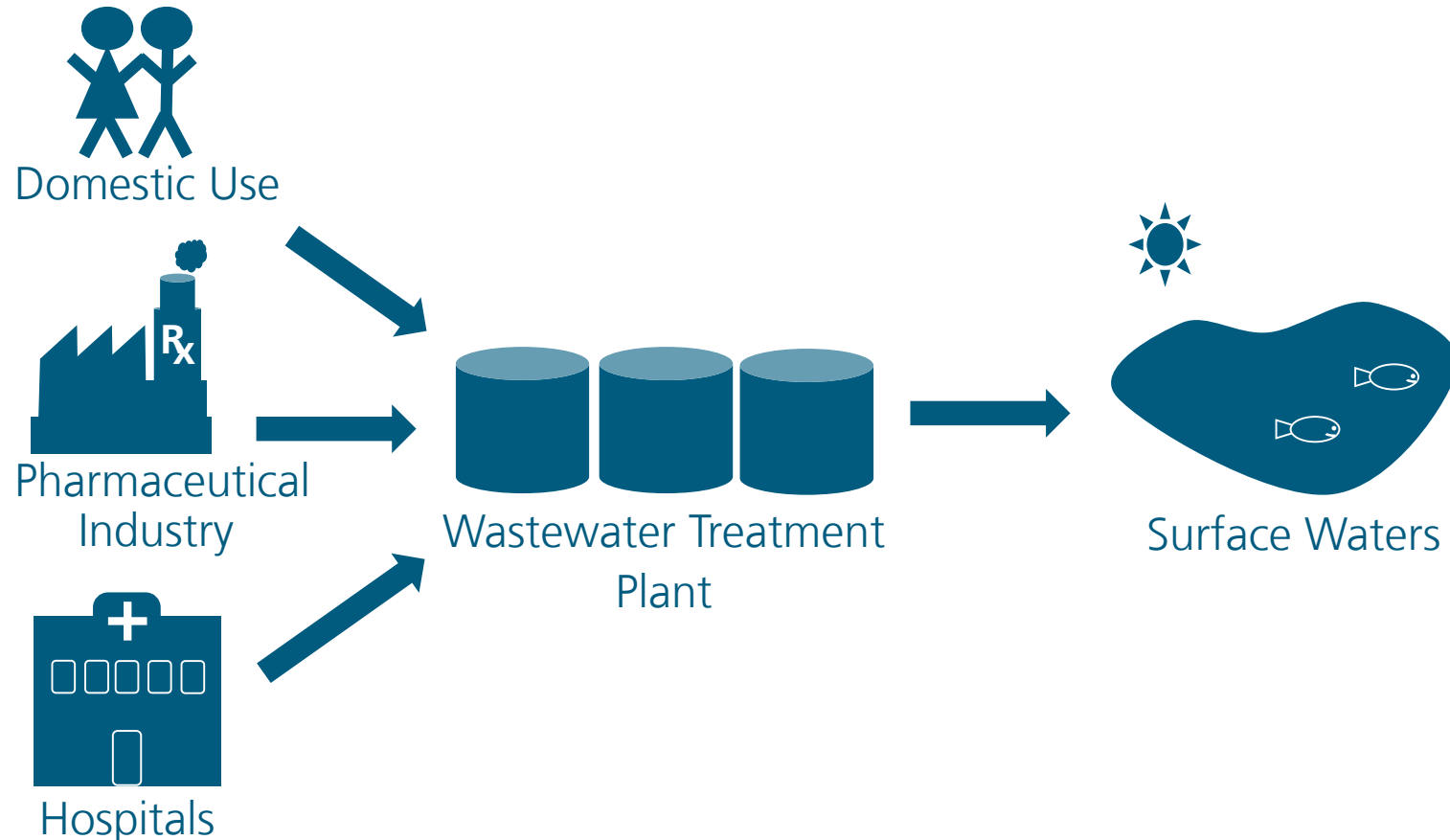
Lotte Ligaya Schaap, Tobias Teichmann, Andre Poremba, Gösta Mattausch, Simone Schopf,
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AGENDA

- Introduction
- Methodology
- Results
- Work in Progress
- Conclusion and Outlook

Introduction

Pharmaceuticals in the Aquatic Environment



- In Germany alone, **9,600 to 12,800 tons of unused pharmaceuticals** are disposed of or flushed down the toilet yearly [1]
- Their presence in effluents have been linked to **ecological toxicity** [2], **antimicrobial resistance** [3], **endocrine disruption**, and **gradual reduction of fertility** [4,5]

Figure 1. Main sources of pharmaceuticals in the aquatic environment.

Introduction

Pharmaceuticals in the Aquatic Environment

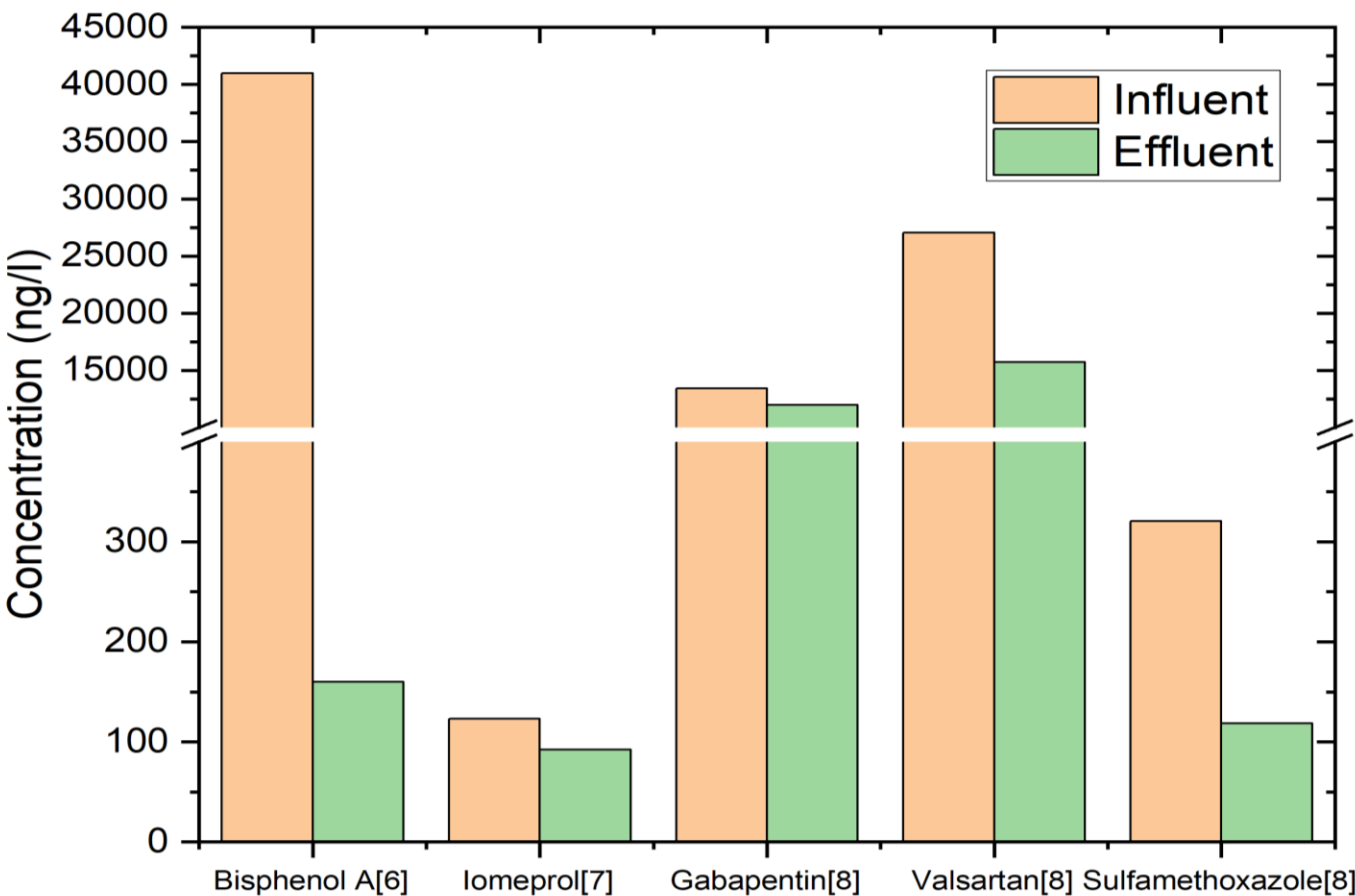
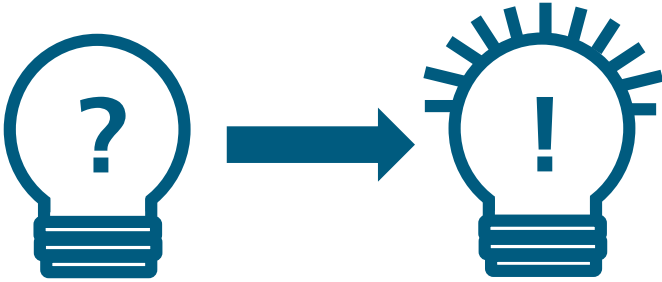


Figure 2. Concentration of micropollutants before and after conventional wastewater treatment.



The Problem:
Pollutants of low-level toxicity which cannot be degraded by traditional methods

The Solution:
Low energy EB Treatment

Introduction

The Problem with Low-Energy Electrons

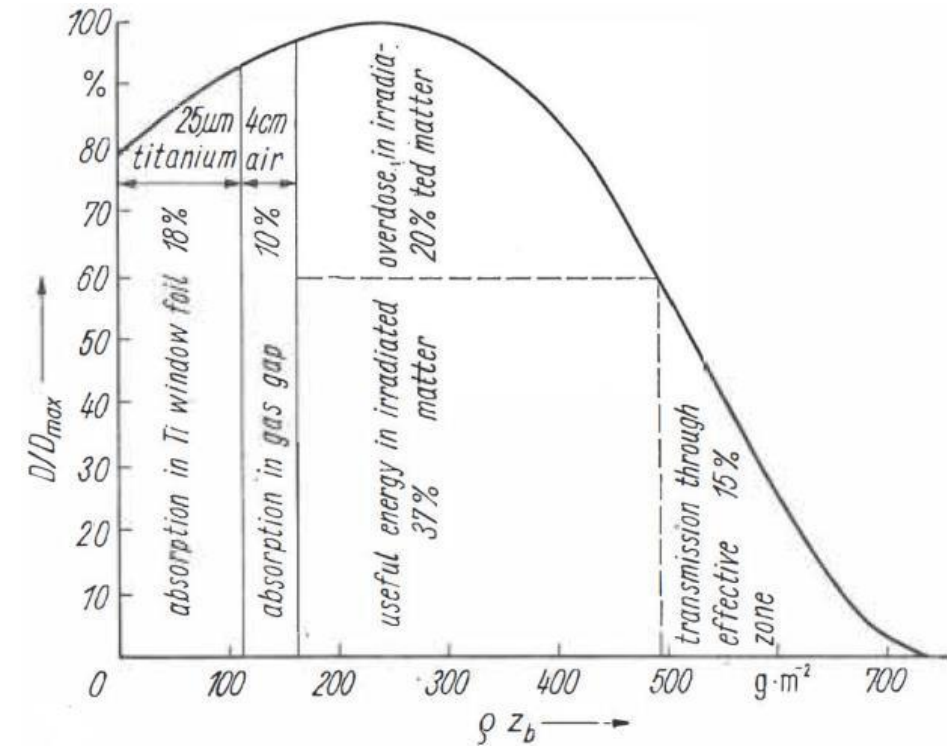
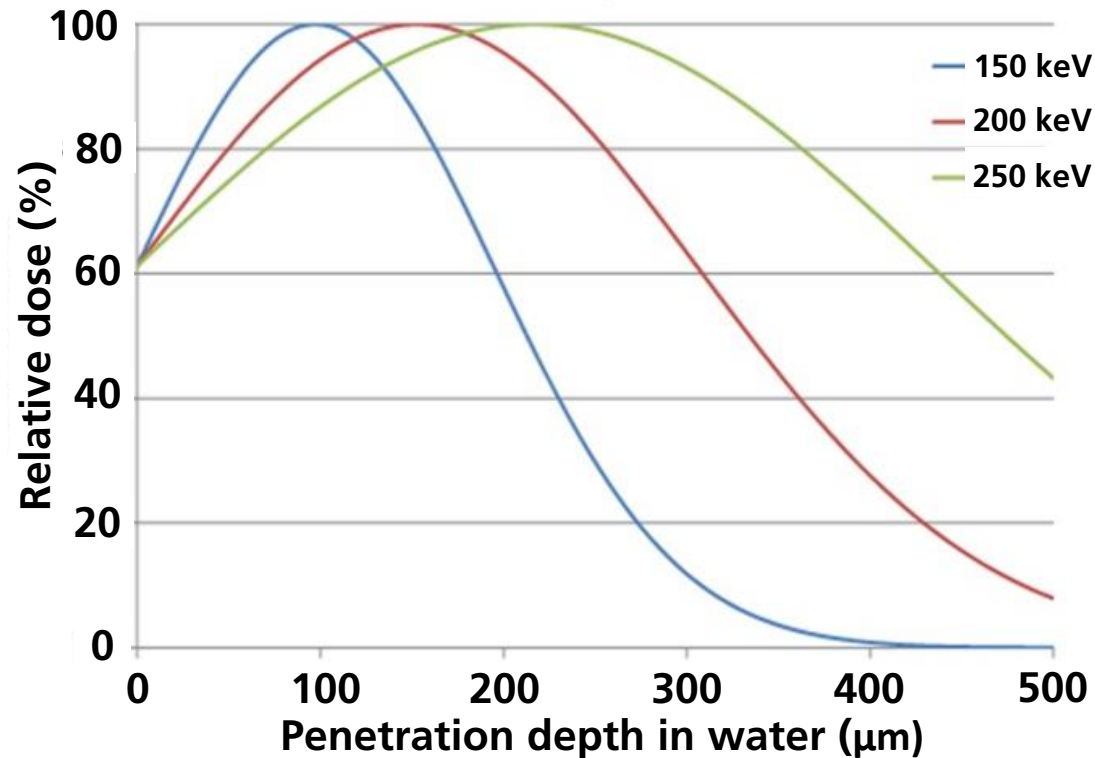
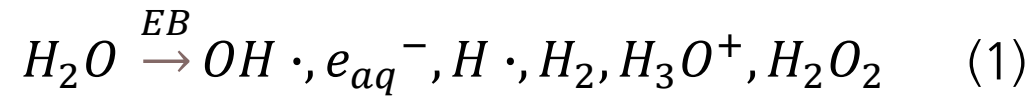


Figure 3. Penetration depth of accelerated electrons according to their energy and density of the absorber (left). Window foil and separating gap are sources of losses in low-energy EB processes (right) [9].

Introduction

Hybrid Wastewater Treatment

The irradiation of water produces several highly reactive, oxidizing and reducing radical species [10]:



By combining low energy EB and ozone:



The demands on radiation protection are kept to a minimum



Low EB source dimensions and costs



An economic, flexible and compact treatment module is realized

Methodology

Experimental Device

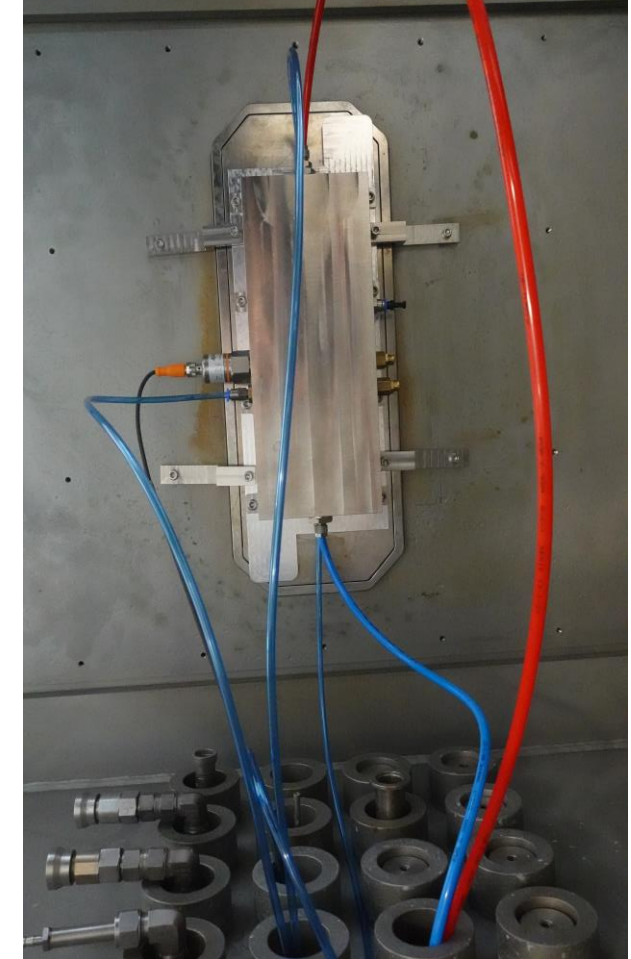
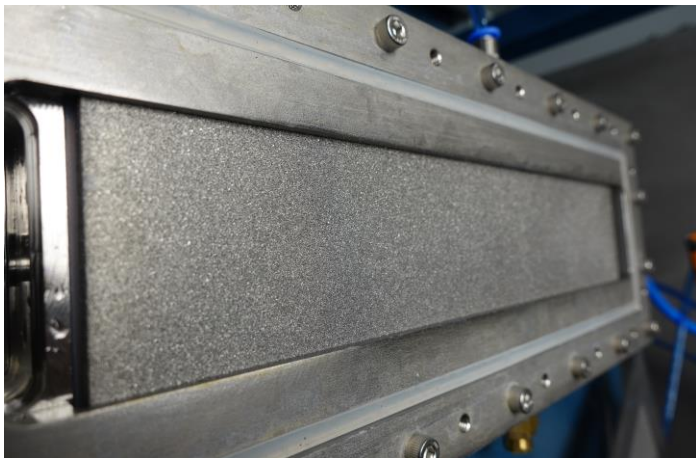
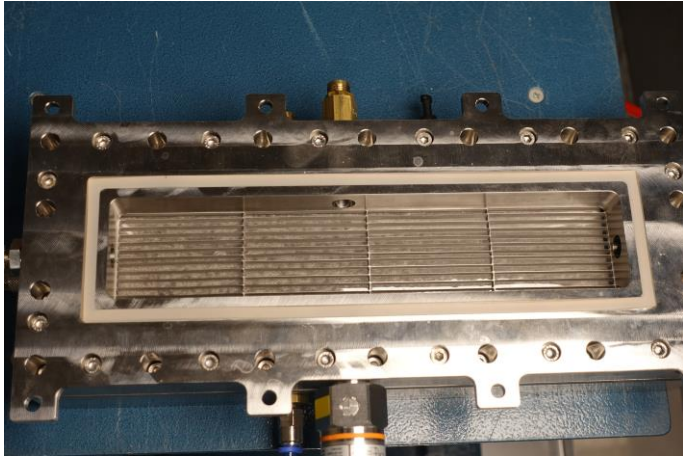
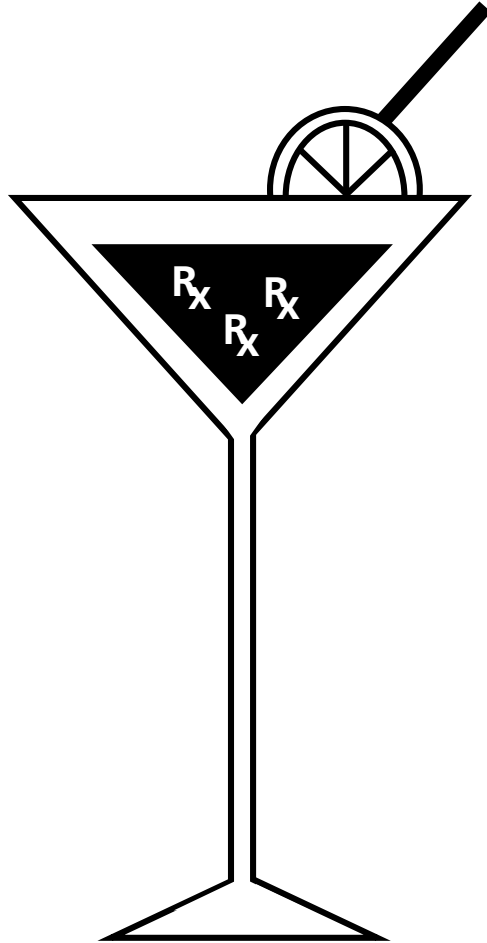


Figure 4. Treatment module (left) and treatment module inside REAMODE (right).

Methodology

Substance Irradiation



Pharmaceutical “cocktail” of five select substances in *deionized (DI) water* and *synthetic wastewater* were mixed



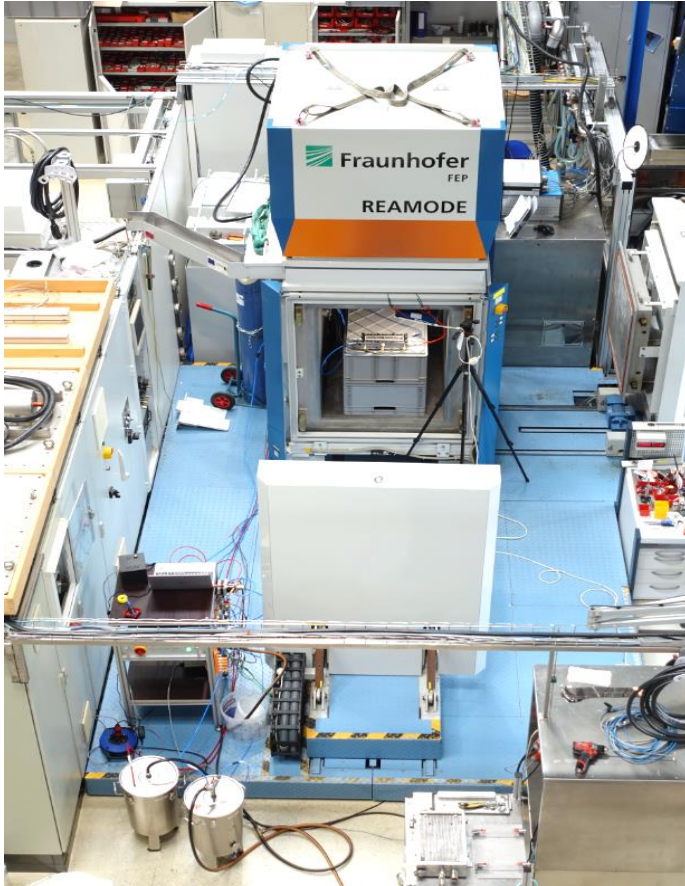
Cocktail was treated using treatment module



Concentration of substances before and after treatment determined using high-performance liquid chromatography

Methodology

Substance Irradiation



Irradiation done at Fraunhofer FEP's electron beam facility REAMODE with the following parameters:

- Maximum electron energy 200 keV
- Irradiation current 0.1-4 mA and 2 mA (multiple passes)
- Volume flow 4 l/min
- Gas flow 2 l/min

Figure 5. EB Facility REAMODE.

Results

Substance Irradiation

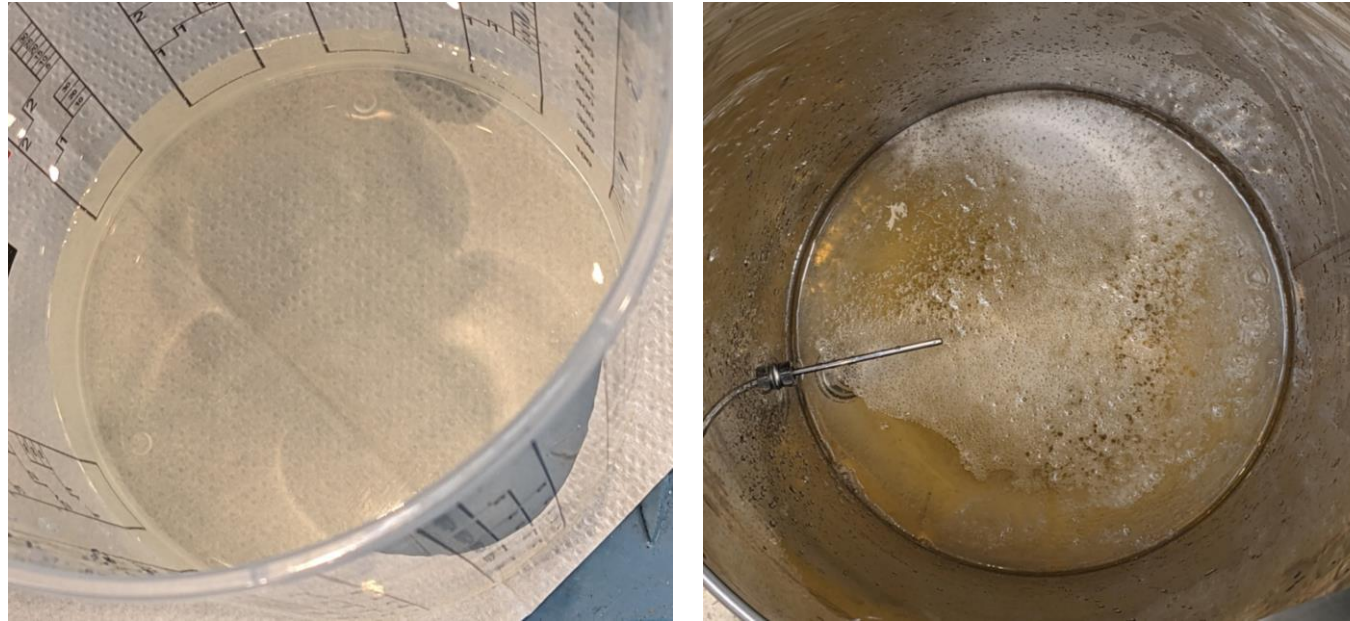
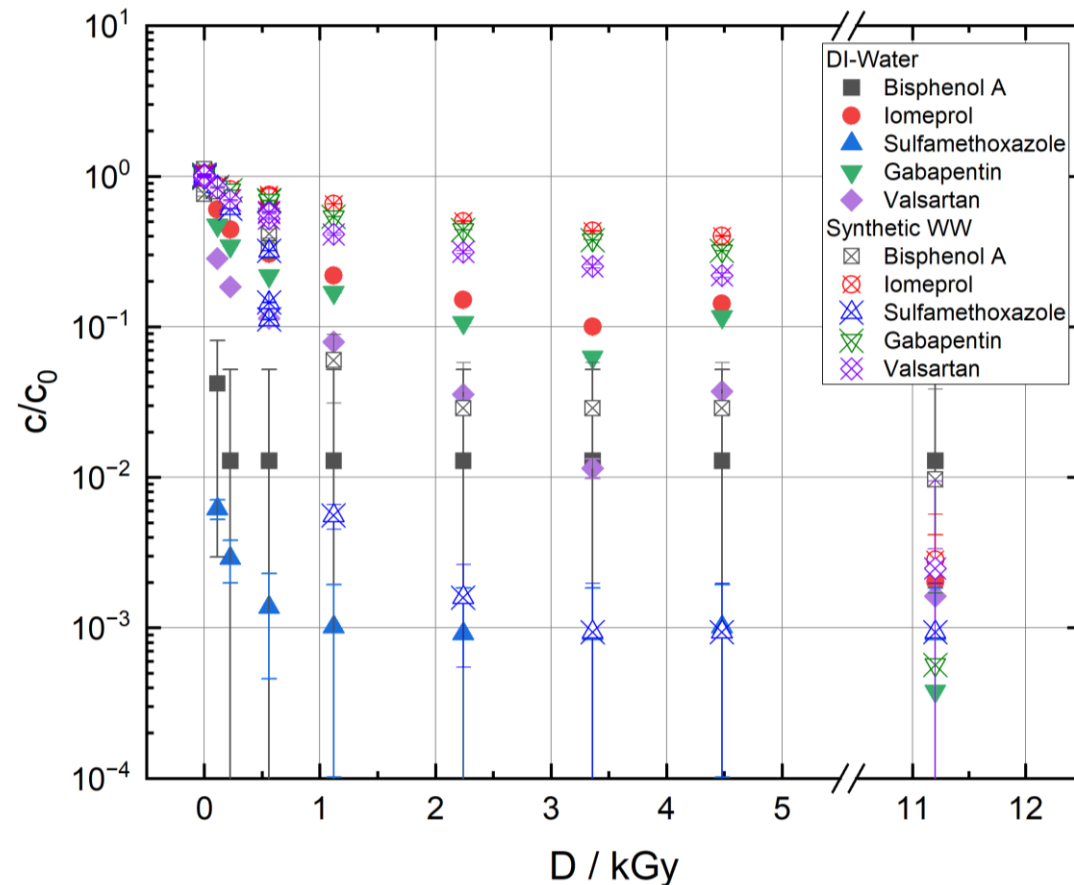


Figure 6. Pharmaceutical cocktail after irradiation.

Results

Substance Irradiation



Significant degradation (>80%) achieved already at 2.2 kGy for all substances in deionized water and synthetic wastewater.

Figure 7. Relative concentration vs dose for five substances in DI-water and synthetic wastewater.

Results

Substance Irradiation

Table 1. Doses at which significant degradation (>80%) is achieved for five select substances in deionised water (DI) and synthetic wastewater (SW).

Substance	Deionised Water	Synthetic Wastewater
Bisphenol A	0.1 kGy	0.6 kGy
Sulfamethoxazole	0.1 kGy	0.6 kGy
Valsartan	0.2 kGy	1.1 kGy
Gabapentin	1.1 kGy	1.1 kGy
lomeprol	2.2 kGy	2.2 kGy

Results

Comparing Wastewater Treatment Methods

Energy Demands of Different Wastewater Treatment Methods

Powdered Organic Carbon (PAC)

10-40 mg/l

- 0.020-0.035 kWh/m³ for adsorption step [11]
- 0.45 kWh/m³ for whole process [12]

Ultraviolet (UV) Treatment

7200 J/m²

- 0.026 kWh/m³ for irradiation step [13]
- 0.5-1 kWh/m³ for whole process [12]

EB + Ozone

2.2 kGy

- 0.003 kWh/m³ for irradiation step

Results

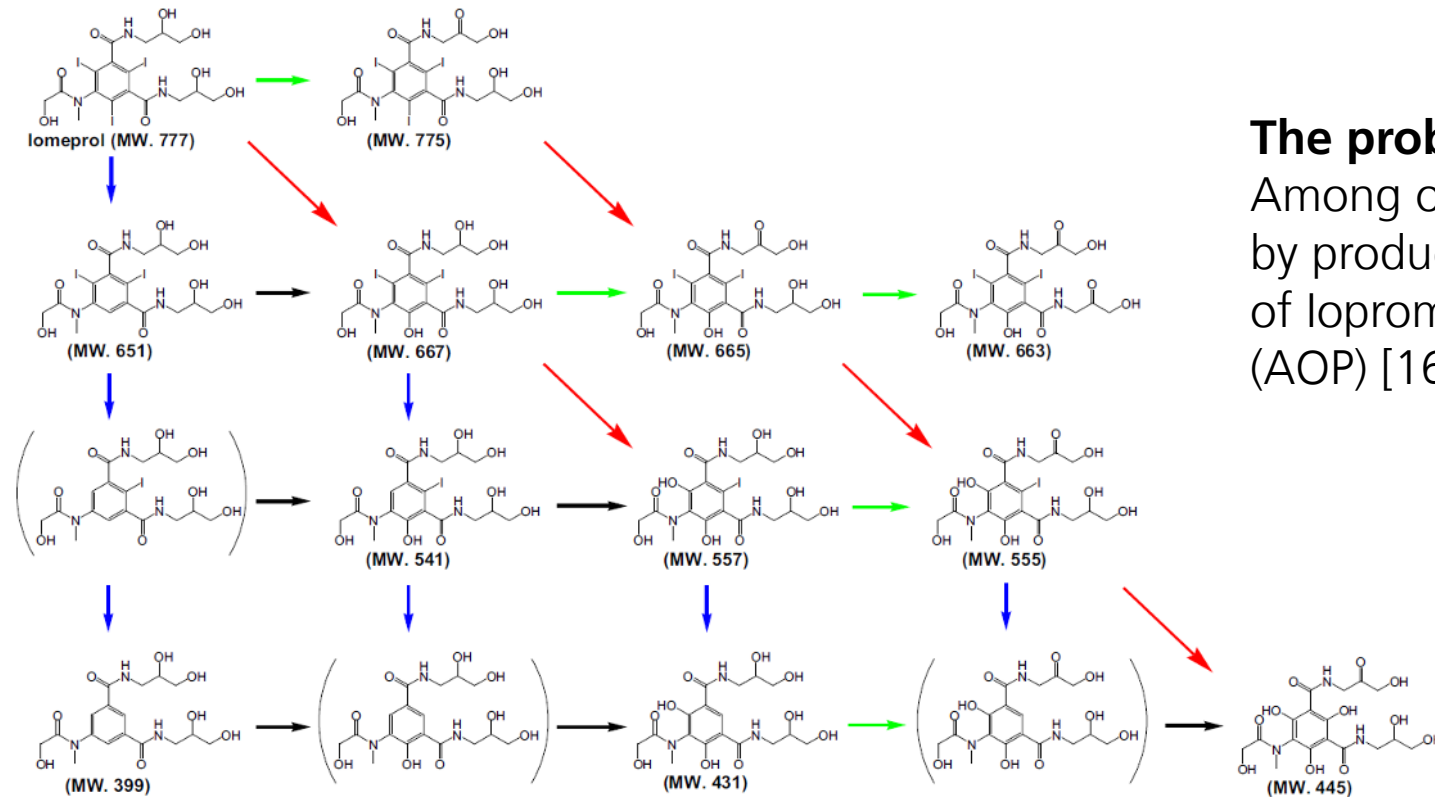
Comparing Wastewater Treatment Methods

Table 2. Measured removal efficiencies with EB and ozone for 5 select substances in synthetic wastewater compared to traditional methods: powdered organic carbon (PAC) and ultra-violet (UV).

Substance	EB + O ₃ (2.2 kGy)	EB + O ₃ (11 kGy)	PAC [14]	UV [14]
Bisphenol A	97.1%	> 99.0%	5.3%	67.0%
Iomeprol	49.7%	> 99.7%	65.0%	65.0%
Sulfamethoxazole	99.8%	> 99.9%	31.0%	81.0%
Gabapentin	56.0%	> 99.9%	41.0%	2.0%
Valsartan	68.6%	> 99.8%	99.0%	5.0%

Work in Progress

Testing Biodegradability



The problem with degradation by-products:
Among other examples, 30 different degradation by products are formed from the transformation of lopromide by advanced oxidation processes (AOP) [16].

Figure 8. Degradation pathway of lopromide. Taken from Jeong et al. [15].

Work in Progress

Testing Biodegradability

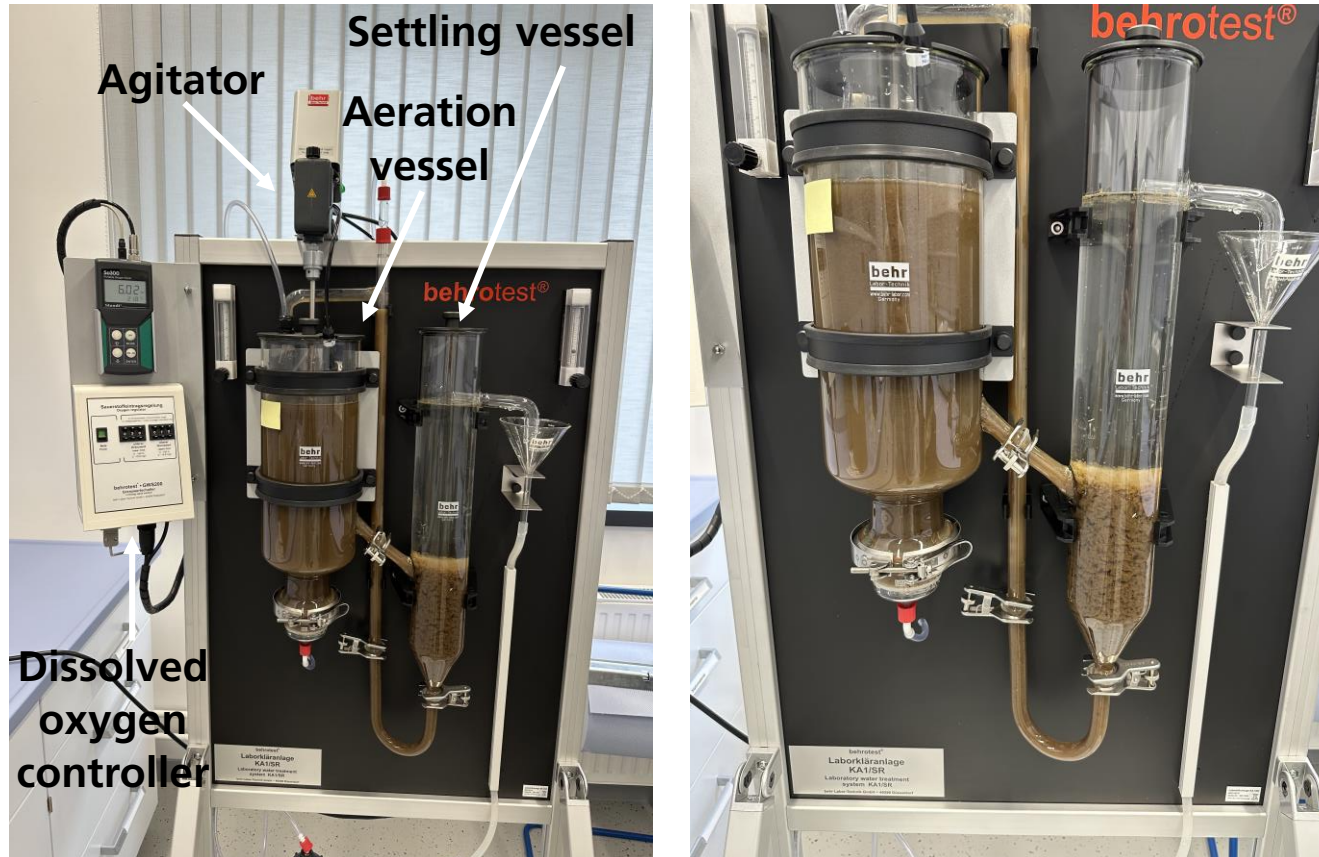


Figure 9. Laboratory scale wastewater treatment plant.

Meet Daisy, FEP's Laborkläranlage!

- The presence of numerous degradation by-products highlights the need for a biodegradability test
- Degradability is measured by taking chemical oxygen demand (COD) measurements
- 70% degradation = inherent ultimate biodegradability

Conclusion and Outlook



The hybrid treatment of EB + Ozone successfully degraded five select pharmaceuticals in DI water and synthetic wastewater



Analysis of degradation byproducts and their fate in further wastewater treatment and the environment is needed



An upscaling of the concept and expansion of its field of application must be developed

Thank you for your attention !



Fraunhofer Institute for Electron
Beam and Plasma Technology FEP

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Acknowledgements

This research was performed in a joint project with
CREAVAC GmbH and co-funded by the European Union



Co-funded by
the European Union



This project is co-financed from tax revenues
on the basis of the budget adopted by the
Saxon State Parliament.

Appendix

Table 3. Removal efficiencies for 5 select substances in deionised water (DI) and synthetic wastewater (SW).

Substance	Medium	RE (2,2 kGy)	RE (11 kGy)
Bisphenol A	DI	> 98,7%	> 98,7%
	SW	97,1%	> 99,0%
Iomeprol	DI	84,8%	> 99,8%
	SW	49,7%	> 99,7%
Sulfamethoxazole	DI	> 99,9%	> 99,9%
	SW	99,8%	> 99,9%
Gabapentin	DI	89,3%	> 99,9%
	SW	56,0%	> 99,9%
Valsartan	DI	96,4%	> 99,8%
	SW	68,6%	> 99,8%

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